

A NEW SYNTHESIS OF 2-SUBSTITUTED BENZOFURANS AND BENZOTHIOPHENS:
 NOVEL FRAGMENTATION REACTIONS OF SIMPLE ALKYL GROUPS

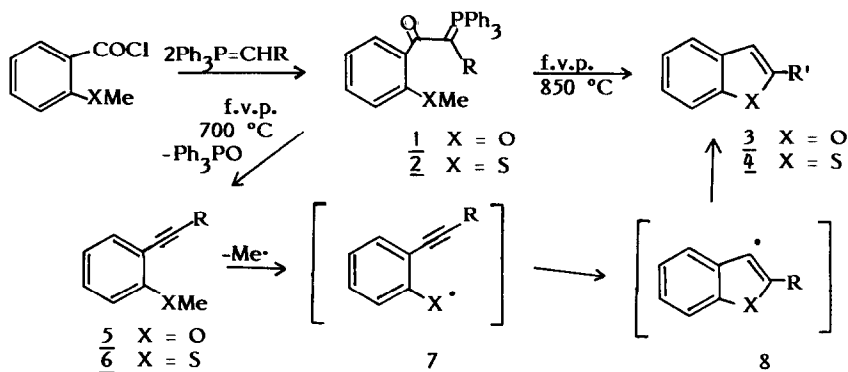
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Summary: Flash vacuum pyrolysis of (2-methoxy- and (2-methylthiobenzoyl)alkylidetriphenylphosphoranes results in loss of Ph_3PO and methyl radical; cyclisation of the resulting radical intermediates leads to benzofurans and benzothiophens in which the 2-substituents have undergone novel fragmentation processes.

A large number of substituted benzofurans possess useful biological activity¹ and their synthesis is of considerable interest². It was recently reported³ that attempted demethylation of (2-methoxyphenyl)alkynes 5 with lithium iodide in 2,4,6-trimethylpyridine led to cyclisation to produce 2-substituted benzofurans. Our recent work on the pyrolysis of β -oxoalkylidetriphenylphosphoranes⁴ suggested not only that this method could provide convenient access to 5, but that, in view of the known thermal lability of aromatic methoxy groups⁵, it might be possible to obtain the benzofuran directly upon pyrolysis of an appropriate β -oxoylide. We now report the success of this strategy which represents a new synthesis of 2-substituted benzofurans and benzothiophens.

The required β -oxoylides 1 and 2⁶ were readily obtained as stable solids by reaction of 2-methoxy- or 2-methylthio-benzoyl chloride with two equivalents⁷ of Wittig reagent. When these were subjected to flash vacuum pyrolysis (f.v.p.) at 700 °C, triphenylphosphine oxide was cleanly eliminated to give the expected alkynes 5 and 6 in good yield. For both $\text{X} = \text{O}$ and S , simply raising the furnace temperature to 850 °C led to complete loss of methyl radical to give the desired heterocycles.



For R = Ph this was straightforward and gave (3, R' = Ph) in 80% and (4, R' = Ph) in 53% yield.

For aliphatic groups R, f.v.p. of 1 and 2 at 850 °C again gave 3 and 4 but the cyclisation was now accompanied by a series of remarkable fragmentation processes. In all the cases examined the original R group had been transformed to give products with R' = Me, Et or vinyl (Table).

FLASH VACUUM PYROLYSIS OF YLIDES 1 TO GIVE 3 AND 2 TO GIVE 4

R	R' =	Yield of <u>3</u> (%)			R' =	Yield of <u>4</u> (%)		
		Me	Et	vinyl		Me	Et	vinyl
Me		8	20	-		14	17	-
Et ⁱ		-	-	83		-	-	49
Pr ⁱ		-	-	60		-	-	24
Pr ⁿ		9	52	10		6	45	9
Bu ⁿ		4	30	25		10	31	23
Pent ⁿ		3	14	21		10	28	14

The formation of 5 and 6 with R unchanged at 700 °C, together with the production of 3 and 4 with R' = Me, Et as stable products at 850 °C, implies that the transformation of the R group occurs at the stage of the radical intermediates 7 and 8. Previous studies on the pyrolysis of related systems have revealed that migration of R in 5 to give a vinylidene carbene is a possibility, leading in the case of 2-alkynylphenols, for example, to 3-substituted benzofurans.^{8,9} However this mechanism does not operate here since all the products obtained had a 3-H substituent and the radical reaction appears to be preferred.

The result for (1, R = Me) is in good agreement with the work of Barton⁹ who obtained (R = Me and Et) in yields of 29% and 15% respectively by pyrolysis of (5, R = Me) at 700 °C and proposed a radical mechanism to account for incorporation of the methoxy carbon atom to form the ethyl group.

The exclusive formation of vinyl compounds in good yield from 1 and 2 with R = Et, Prⁱ is readily explained by intramolecular abstraction of a β-hydrogen atom in 8 via a five-membered transition state with expulsion of H· or Me·. Increasing the length of the carbon chain in R gave more complex product mixtures but the total yields of products 3 and 4 were still reasonable. For R = Prⁿ these were the only volatile products and for R = Buⁿ and Pentⁿ no other benzofurans or benzothiophens were detected. The mechanisms for these interesting and novel transformations of simple alkyl groups are currently under investigation.

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REFERENCES and NOTES

1. P. Cagniant and D. Cagniant, *Adv. Heterocycl. Chem.*, 1975, **18**, 337.
2. See for example: B. Ledoussal, A. Gorgues, and A. Le Coq, *J. Chem. Soc., Chem. Commun.*, 1986, 171; J. Grimshaw and N. Thompson, *ibid.*, 1987, 240.
3. D. R. Buckle and C. J. M. Rockell, *J. Chem. Soc., Perkin Trans.1*, 1985, 2443.
4. R. A. Aitken and J. I. Atherton, *J. Chem. Soc., Chem. Commun.*, 1985, 1140.
5. L. K. Freidlin, A. A. Balandin and N. M. Nazarova, *Izv. Akad. Nauk SSSR, Otdel. Khim. Nauk*, 1949, 102.
6. All new compounds gave satisfactory spectroscopic and microanalytical data.
7. H. J. Bestmann and B. Arnason, *Chem. Ber.*, 1962, **95**, 1513.
8. R. Bloch and P. Orvane, *Tetrahedron Lett.*, 1981, **22**, 3597.
9. T. J. Barton and B. L. Groh, *J. Org. Chem.*, 1985, **50**, 158.

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